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Key indicators

Single-crystal X-ray study T = 295 KMean $\sigma(\text{C}-\text{C}) = 0.007 \text{ Å}$ Disorder in main residue R factor = 0.051 wR factor = 0.110 Data-to-parameter ratio = 15.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

(2-Hydroxy-3,5-dinitrobenzoato)tris(2-methyl-2-phenylpropyl)tin(IV)

The Sn atom of the title compound, $[Sn(C_{10}H_{13})_3(C_7H_3N_2O_7)]$, is four-coordinate and possesses a distorted tetrahedral geometry. There is an intramolecular $O-H\cdots O$ hydrogen bond between the carboxyl and the phenolic hydroxy groups.

Comment

Recently, we have reported several structures of tris(2-methyl-2-phenylpropyl)tin carboxylates, such as tris(2-methyl-2phenylpropyl)tin pyridine-3-carboxylate (Tian, Sun, Yang & Yang, 2005), 3,5-dinitrobenzoate (Tian, Yu, Sun, & Zhang, 2005) and bis[tris(2-methyl-2-phenylpropyl)tin(IV)] phthalate (Tian, Sun, Yang & Ng, 2005), which all possess a distorted tetrahedral geometry. In the title compound, (I), tetrahedral coordination is also observed (Fig. 1).



The Sn1···O2 separation of 3.118 (3) Å indicates a weak interaction between these atoms, which distorts the tetrahedral geometry. Bond dimensions around the Sn atom (Table 1) are similar to those found in the carboxylate structures mentioned above. An intramolecular $O-H\cdotsO$ hydrogen bond is observed between the carboxyl and the phenolic hydroxy groups (Fig. 1 and Table 2). The OH group is disordered over two *ortho* positions, and both O atoms of the carboxylate functionality may serve as acceptor for the intramolecular hydrogen bonding, giving hydrogen bonds of very similar strengths. An unbalanced statistical distribution, 0.836 (7)/0.164 (7), clearly reflects the fact that atom O2 is a better acceptor than O1.

Experimental

Bis[tris(2-methyl-2-phenylpropyl)tin] oxide (1.05 g, 1 mmol) and 2hydroxy-3,5-dinitrobenzoic acid (0.46 g, 2 mmol) in toluene (50 ml) were refluxed for 5 h with azeotropic removal of water *via* a Dean– Stark trap. The resulting clear solution was evaporated under reduced pressure. The white solid obtained, (I), was purified by recrystal-

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lization from ethanol and crystals were obtained from a chloroformhexane (1:1 ν/ν) solution by slow evaporation at 298 K (yield 73%, m.p. 382–383 K). Analysis found: C 59.76, H 5.59, N 3.65%; calculated for C₃₇H₄₂N₂O₇Sn: C 59.61, H 5.68, N 3.76%. IR (KBr disc): ν_{as} (CO₂) 1651, ν_{s} (CO₂) 1340, ν_{as} (NO₂) 1536, ν_{s} (NO₂) 1398 cm⁻¹; ¹H NMR (400 MHz, CDCl₃, TMS as reference): δ 9.31 (2H, *s*, nitrobenzene H), 7.28–7.05 (15H, *m*, 3× C₆H₅), 5.17 (1H, *s*, OH), 1.30 (6H, *s*, 3 × CH₂Sn), 1.24 (18H, s, 6 × CH₃); ¹¹⁹Sn NMR (111.9 MHz, CDCl₃, Me₄Sn as reference): δ 112.3.

Crystal data

| $D_x = 1.382 \text{ Mg m}^{-3}$ |
|---|
| Mo $K\alpha$ radiation |
| Cell parameters from 7082 |
| reflections |
| $\theta = 2.3 - 25.6^{\circ}$ |
| $\mu = 0.76 \text{ mm}^{-1}$ |
| T = 295 (2) K |
| Needle, pale yellow |
| $0.25 \times 0.10 \times 0.08 \text{ mm}$ |
| |

6649 independent reflections 5673 reflections with $I > 2\sigma(I)$

 $R_{\rm int} = 0.040$

 $\theta_{\rm max} = 25.5^{\circ}$ $h = -12 \rightarrow 12$

 $k = -27 \rightarrow 26$

 $l = -19 \rightarrow 19$

Data collection

| Bruker APEX area-detector |
|--|
| diffractometer |
| φ and ω scans |
| Absorption correction: multi-scan |
| (SADABS; Bruker, 2002) |
| $T_{\min} = 0.832, \ T_{\max} = 0.942$ |
| 26672 measured reflections |

Refinement

| Refinement on F^2 | $w = 1/[\sigma^2(F_0^2) + (0.0384P)^2]$ |
|---------------------------------|--|
| $R[F^2 > 2\sigma(F^2)] = 0.051$ | + 1.8962P] |
| $wR(F^2) = 0.110$ | where $P = (F_0^2 + 2F_c^2)/3$ |
| S = 1.18 | $(\Delta/\sigma)_{\rm max} = 0.001$ |
| 6649 reflections | $\Delta \rho_{\rm max} = 0.91 \text{ e } \text{\AA}^{-3}$ |
| 434 parameters | $\Delta \rho_{\rm min} = -0.79 \text{ e} \text{ \AA}^{-3}$ |
| H-atom parameters constrained | |

Table 1

Selected geometric parameters (Å, °).

| Sn1-O1 | 2.109 (3) | O2-C1 | 1.242 (5) |
|------------|-------------|-------------|-------------|
| Sn1-C18 | 2.144 (4) | O7A - C3 | 1.326 (5) |
| Sn1-C8 | 2.146 (4) | O7B-C7 | 1.393 (17) |
| Sn1-C28 | 2.147 (4) | C1-C2 | 1.479 (5) |
| O1-C1 | 1.273 (5) | | |
| O1-Sn1-C18 | 104.07 (14) | O1-Sn1-C28 | 90.41 (15) |
| O1-Sn1-C8 | 101.84 (15) | C18-Sn1-C28 | 118.70 (17) |
| C18-Sn1-C8 | 116.22 (16) | C8-Sn1-C28 | 118.25 (17) |

Table 2

Hydrogen-bond geometry (Å, °).

| $D - H \cdot \cdot \cdot A$ | D-H | $H \cdot \cdot \cdot A$ | $D \cdots A$ | $D - H \cdots A$ |
|--|--------------|-------------------------|-------------------------|------------------|
| $O7B-H7BA\cdots O1$ $O7A-H7AA\cdots O2$ | 0.82 0.82 | 1.74 1.73 | 2.465 (18) 2.477 (4) | 147 150 |



Figure 1

The structure of (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level. The disordered hydroxy group (O7a/O7b) is represented with different bond styles. Except for the OH groups, H atoms have been omitted. Hydrogen bonds are shown as dashed lines.

The hydroxyl group of the hydroxydinitrobenzoate ligand is disordered over two sites. Site-occupation factors were refined with a sum constrained to 1, converging to 0.836 (7) for atom O7*a* and 0.164 (7) for O7*b*. H atoms were placed at calculated positions and were included in the refinement in the riding-model approximation $[C-H = 0.93 \text{ Å} \text{ and } U_{iso}(H) = 1.2U_{eq}(C)$ for aromatic H atoms; $C-H = 0.96 \text{ Å} \text{ and } U_{iso}(H) = 1.5U_{eq}(C)$ for methyl H atoms; C-H = 0.97 Å and $U_{iso}(H) = 1.2U_{eq}(C)$ for methylene H atoms; O-H = 0.82 Å and $U_{iso}(H) = 1.5U_{eq}(O)$ for hydroxy H atoms].

Data collection: *SMART* (Bruker, 2002); cell refinement: *SAINT* (Bruker, 2002); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP3* for Windows (Farrugia, 1997); software used to prepare material for publication: *SHELXL97* (Bruker, 2002).

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References

- Bruker (2002). SADABS, SAINT and SMART. Bruker AXS Inc., Madison, Wisconsin, USA.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Tian, L.-J., Sun, Y.-X., Yang, M. & Ng, S. W. (2005). Acta Cryst. E61, m74-m75. Tian, L.-J., Sun, Y.-X., Yang, M. & Yang, G.-M. (2005). Acta Cryst. E61,
- m1346–m1347.
- Tian, L.-J., Yu, F.-Y, Sun, Y.-X. & Zhang, B. (2005). Acta Cryst. E61, m1537– m1538.